UNIVERSITY OF MUMBAI



Syllabus for Sem V & VI
Applied Component

Program: B.Sc.

Course: DRUGS & DYES

(Credit Based Semester and Grading System with effect from the academic year 2013–2014)

T.Y.B.Sc. Applied Component DRUGS & DYES Syllabus

Credit Based Semester and Grading System To be implemented from the Academic year 2013-2013

SEMESTER V

Theory

COURSE: USACDD501

UNIT	TOPICS	CREDIT	L/WEEK
I	1.1 General Introduction to Drugs 1.1 .1 Definition of a drug, Requirements of an ideal drug, Classification of drugs (based on therapeutic action) 1.1.2 Nomenclature of drugs: Generic name, Brand name, Systematic name 1.1.3 Definition of the following medicinal terms: Pharmacon, Pharmacophore, Prodrug, Half-life efficiency, LD ₅₀ , ED ₅₀ , Therapeutic Index. 1.1.3 Brief idea of the following terms: Receptors, Drug-receptor interaction, Drug Potency, Bioavailability, Drug toxicity, Drug addiction, Spurious Drugs, Misbranded Drugs, Adulterated Drugs, Pharmacopoeia. 1.2. Routes of Drug Administration and Dosage Froms (2L) 1.2.1 Oral and Parenteral routes with advantages and disadvantages. 1.2.2 Formulations, Different dosage forms(emphasis on sustained release formulations.) 1.3. Pharmacodynamic agents A brief introduction of the following pharmacodynamic agents and the study with respect to their chemical structure, chemical class, therapeutic uses, and side effects. 1.3.1 CNS Drugs: (5L) Classification based on pharmacological actions Concept of sedation and hypnosis, anaesthesia. Phenobarbitone (Barbiturates), Phenytoin	2	4

(Hydantoins), Trimethadione (Oxazolidinediones), Piracetam (Pyranones), Midazolam, Alprazolam (Benzodiazepines) Methylphenidate (Piperidines) Chlorpromazine (Phenothiazines) Fluoxetine (Phenyl propyl amines) Synthesis of Trimethadione, Methylphenidate, Phenytoin. 1.3.2 Analgesics and Antipoyretics (2L) Morphine (Phenanthrene alkaloids), Tramadol (Cyclohexanols), Aspirin (Salicylates), Paracetamol (p-Aminophenols), Synthesis of Tramadol, Paracetamol. **2.1 Anti-inflammatory Drugs** (2L) Mechanism of inflammation and various inflammatory conditions. Prednisolone, Betamethasone (Steroids), Aceclofenac (N-Aryl anthranilic acids), Mefanic Acid (N-Aryl anthranilic acids). Synthesis of Aceclofenac **2.2** Antihistaminic Drugs (2L) Mechanism of histamine release & its action Diphenhydramine (ethanolamines), Cetrizene (piperazine), Chloropheniramine maleate (ethyl amines), Omeprazol, pantoprazol (Benzimidazoles) Synthesis of cetrizine 2.3 Cardiovascular drugs (3L) Classification based on pharmacological action Enalapril (a-amino acids), Isosorbide dinitrate II (Nitrates), Atenoldol (Aryloxy propanol amines), Nifedipine (Pyridines), Chlorthiazide (Thiazides), Frusemide /Furosemide (Sulfamyl benzoic acid), Spironolactone (Steroidal-17-Y-lactones). Synthesis of Furosemide, Atenolol from 3-Hydroxy phenyl acetamide 2.3 Antidiabetic Agents (2L)General idea and types of diabetes; Insulin therapy Glibenclamide (sulphonyl ureas), Metformin (Biguanides), 2.5 Antiparkinsonism Drugs(2L) Idea of Parkinson's disease. Procyclidine hydrochloride (Pyrrolidines), Ethopropazine hydrochloride (Phenothiazines) Laevodopa (a-amino acids) Synthesis of Levodopa from Vanillin. 2.6 Drugs for Respiratory System (2L)

	Compared idea of Expressions to Marcolate		
	General idea of Expectorants; Mucolytes; Bronchodilators		
	Decongestants and Antitussives, Bromhexine		
	(Phenyl methyl amines), Salbutamol, Pseudo-		
	ephedrine (Phenyl ethyl amines)		
	Oxymetazoline (Imidazolines)		
	Codeine Phosphate (Opiates)		
	Synthesis of Salbutamol		
	2.7 Mode of Action of the Following Drugs (2L)		
	Barbiturates (As sedatives and hypnotics), Atenolol		
	(As β-1 blocker), Diphenhydramine (As		
	Anthistaminic agent), Glibenclamide (As oral		
	hypoglycemic agent)		
	3.1 Introduction to Dyestuff Chemistry (5L)		
	3.1.1 Important landmark in the history of dyes		
	3.1.1.1 Natural colouring matter and their		
	limitations:e.g,; Heena, Turmeric, kesar,		
	Chlorolphyll, Indigo, Alizarine from roots of madder		
	plants, Logwood. Tyrian		
	Purple.		
	3.1.1.2 Synthetic Dyes: Important molestones, i.e.		
	Mauve, Diazotization, aniline Yellow, Congo Red,		
	Synthesis and structure of Indigo, disperse Dye,		
	fluorescent Brighteners, procion reactive Dyes,		
	Remazole Dyes. (Emphasis on Name of the Scientist		
	and dyes and the year of		
	the discovery is required and structure is not		
	expected		
	3.1.2 Defination of dyes, Properties i.e. colour,		
III	Chromophore and Auxochrome, Solubility, Linearity,		
	Coplanarity, fastness properties, substantivity,		
	Economic viability		
	3.1.3 Explanation of nomenclature of commercial		
	dyes with atleast one example .suffixes-G, O, R, B,		
	6B, GK, 3GK, 6GK, L, S Explanation: naming of		
	dyes by colour index(two examples)		
	3.2. Classification of dyes based on constitution		
	(3L)		
	(Examples are mention below with structures)		
	(i) Nitro Dyes-Napyhol yellow S		
	(ii) Nitroso Dye-Gambine Y		
	(iii) Azo Dyes- (a) Monoazo Dyes- Metanil yellow		
	(b) DiazoDyes- Napthol Blue Black (c) Triazodyes -		
	Chloroamine Green B		
	(iv) Diphenymethane Dyes-Auramine G		
	(v) Triphenyl methane Dyes-		

(a) Malachite Green Series- Naphthalene green V (b) Magenta Series- Acid Magenta (c) Rosolic acid Series-Chrome Violet (vi) Heterocyclic Dyes (a) Xanthene-Rhodamine 6G (b) Acridines-Acriflavine (c) Azines- SafranineB (d) Oxazines-Capri blue (e) Thiazines-Methylene Green (f) Ouiolines- Ouinoline Yellow (g) Thiazoles-Primuline (vii) Benzoquinones and naphthaguinones-Napthazarin (viii) Anthraquinone Dyes- Indanthrene, Turquoise Blue 3GK (ix) Indigoids-Indigo Caramine (x) Pthacyanines-Sirius Light green FFGL 3.3 Classification Based on Application (6L) Definition, fastness properties & applicability on substrates examples with structures (a) Acid Dyes- Orange II, (b) Basic Dyes-methyl violet, Victoria Blue B (c) Direct cotton Dyes-Benzofast Yellow 5GL (d) Azoic Dyes-Diazo components; Fast yellow G,Fast orange R. Coupling components. Naphtol AS, Naphthol ASG (e) Mordant Dyes-Erichrome Black A, Alizarin. (f) Vat Dyes- Indanthrene brown RRD, Indanthrene Red 5GK. (g) Sulphur Dyes- Sulphur Black T (no structure) (h) Disperse Dyes-Celliton Fast brown 3R, perlon fast blue FFR (i) Reactive Dyes- cibacron Brillant Red B, procion brillant Blue HB. 4. 1 Colour and chemical constitution of dyes (5L)4.1 .1 Absorption of visible light, colour of wavelength absorbed, complementary colour. 4.1.2 relation between colour and chemical constitution. (i) Armstrong theory (quinonoid theory) and its IVlimitations (ii) Valence Bond theory; Comparative study and relation of colour in the following classes of compounds/dyes: Benzene, Nitrobenze, Nitroanilines, Nitrophenols, Benzoquinones, Azo, Triphenyl methane, Anthraquinones. (iii) Molecular Orbital Theory.

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	4.2. Non-textile Uses of Dyes (6L)	
	Structural features of the substrate, fastness and other	
	property requirements and main classes of dyes used	
	to be mentioned as applicable. (Two examples with	
	structures for each of the following.) . 1. Leather 2.	
	5 /	
	Paper 3. Foodstuff 3. Cosmetics	
	5. Medicinal 6. Biological Stains 7. Indicator &	
	Analytical Reagents 3. Coloured Smokes &	
	Camouflage colours 3. Laser Dyes	
	4.3 Optical BrightenerS (2L)	
	General idea and important characteristics of optical	
	brighteners, one example each with structure of the	
	following classes: Stilbene, Coumarin, Heterocyclic	
	vinylene derivatives, Diaryl pyrazolines,	
	Naphthalimide derivatives.	
	4.4 Organic Pigments (2L)	
	General idea, distinguish between dyes and pigments,	
	important characteristics of organic pigments,	
	Toners, Lakes, Classification of organic pigments	
	with suitable examples, i.e. Ionic pigments-Lake of	
	acid and basic dyes. Nonionic pigments-Azo,	
	Indigoid, Anthraquinone, Quinacridone,	
	Phthalocyanine (Copper phthalocyanine).	

Practicals

USACHDD5P1

	I) Dye Preparation: 1) Preparation of Orange-II		
	II) Dyes Estimation:1. Estimation of Primary amino group by diazotisation		
USACHDD 5P1	 III) Drug Estimation: 1. Estimation of Ibuprofen 2. Estimation of Acid neutralizing capacity of a drug IV)Preparation of monogram of any one drug from syllabus by I.P. method 	2	4
	V) Drug Preparations: 1) Preparation of p-Nitroacetanilide from Acetanilide		

2) Preparation of p-Nitroaniline from p-Nitroacetanilide3) Preparation of Methyl Salicylate from Salicylic Acid	

SEMESTER VI

Theory

COURSE	UNIT	TOPICS	CREDIT	L/WEEK
USACDD601	I	1.1 Drug Discovery, Design and Development (5L) 1.1.1 Discovery of a Lead compound: Screening, drug metabolism studies and clinical observation. 1.1.2 Drug development from Natural Sources: Anti infective agents Anti cancer agents CNS agent 1.1.3 Development of drug: The Pharmacophore identification, modification of structure or functional group, Structure activity relationship (Benzodiazepines, Sulphonamides). 1.1.3 Structure modification to increase potency: Homologation, Chain branching, Ring-chain transformation, Extension of the structure. 1.1.5 Computer assisted drug design. 1.2 Drug Metabolism (3L) Introduction, Absorption, Distribution, Bio-transformation, Excretion Different types of chemical transformation of drugs with specific examples.	2	4

	1.2 Chamathanar 4 - A 4		
	1.3 Chemotherapeutic Agents		
	(1L)		
	Study of the following		
	chemotherapeutic agents with		
	respect to their chemical		
	structure, chemical class,		
	therapeutic uses, and side effects.		
	1.3.1 Antibiotics (2L)		
	Definition, Characteristics and		
	properties of:		
	Amoxicillin; Cloxicillin (\(\beta \)-		
	lactum antibiotics)		
	Cephalexin (Cephalosporins)		
	Doxycycline (Tetracyclines)		
	Gentamycin (Aminoglycosides)		
	Ciprofloxacin (Quinolones)		
	Synthesis of Ciprofloxacin		
	1.3.2 Antimalarials (2L)		
	Types of malaria: Symptoms;		
	pathological detection during		
	window period (Life cycle of the		
	parasites not o be discussed)		
	Chloroquine (3-Amino quinolines)		
	Paludrine (Biguanides)		
	Pyrimethamine (Diamino		
	pyrimidines)		
	Artemether (Benzodioxepins)		
	Following combination to be		
	discussed		
	(i) Sulfadosine-Pyrimethamine		
	(ii) Atremether-Lumefantrine (no		
	structure)		
	Synthesis of Paludrine.		
	1.3 3 Anthelmintics (2L)		
	. ,		
	Drugs effective in the treatment of		
	Nematodes and Cestodes		
	intestations.		
	Diethyl carbamazine (Piperazines)		
	Mebandazole; Albendazole		
	(Benzimidazoles)		
	Niclosamide (Amides)		
	Synthesis of Albendazole		
	2.1 Antiamoebic Drugs (1L)		
TT	Types of Amoebiasis		
II	Metronidazole;		
	Diloxamide furoate (Furans)		
<u> </u>		l	I .

Following combination therapy to be discussed: Ciprofloxacin-Tinidazo Synthesis of Metronidazole 2.2 Antitubercular and **Antileprotic Drugs** (3L) Types of Tuberculosis; Symptoms and diagnosis of Tubeculosis. Types of Leprosy. General idea of Antibiotics used in their treatment. PAS (Aminosalicylates) Isoniazide (Hydrazides) Pyrazinamide (Pyrazines) (+) Ethambutol (Aliphatic diamines) Ethionamide (Thioamides) Dapsone (Sulfonamides) Clofazimine (Phenazines) Following combination therapy to be discussed: (i) Rifampin + Ethambutol + Pyrazinamide (ii) Rifampin + Isoniazide + Pyrazinamide (iii) Rifampin + Clofazimine + Ethionamide. Synthesis: (+) Ethambutol, Dapsone. 2.3 Anti-Neoplastic Drugs (2L) Idea of malignancy; Causes of cancer, brief idea of Immuno Stimulants, Immuno depressants. (1) Lomoustine (Nitrosoureas) (2) Fluorouracil (Pyrimidines) (3) Estrogen (Steroidal hormones) (3) Mitomycin C (Antibiotics) (5) Vincristine; vinblastine; vindesine (Vica alkaloids-no structures) Synthesis of 5-Fluorouracil from 2.3 **Anti HIV Drugs** (1L) Idea of HIV pathogenecity, Symptoms of AIDS, AZT, Lamivudine, Stavudine

	(Pyrimidines), DDI (Purines)	<u> </u>
	2.5 Drug Intermediates:	
	Synthesis and uses (3L)	
	(i) 2-Amino-5-	
	chlorobenzophenone from p-	
	chloronitrobenzene	
	(ii) 2,3,6-Triamino-6-	
	hydroxypyrimidine from	
	Guanidine.	
	(iii) 3-Chloro-5-sulphonyl amino	
	anthranilic acid from 3-Chloro-2-	
	toludine	
	(iv) p-[2'-(5-Chloro-2-methoxy	
	benzamido) ethyl]-	
	benzenesulphonamide from	
	Methyl-5-chloro-2-	
	methoxybenzene	
	(v) 3-(p-Chlorophenyl)-3-	
	hydroxypiperidine from 3-	
	Chloroacetophenone.	
	(vi) p-Acetyl amino	
	benzenesulphonyl chloride from	
	Aniline	
	(vii) Epichlorohydrine from	
	propene.	
	2.6 Nano particles in Medicinal	
	Chemistry (3L)	
	Introduction, Carbon nano	
	particles (structures), Carbon nano	
	tubes:	
	Functionalisation for	
	Pharmaceutical applications	
	Targeted drug delivery	
	In vaccine (Foot and mouth	
	disease)	
	Use in Bio-physical treatment.	
	Gold nano particles in treatment of	
	cancer, Parkinsonism, Alzheimer.	
	Silver nano particles:	
	Antimicrobial activity.	
	3.1 Intermediates (11L)	1
	3.1 A brief idea of Unit processes	
	3.1.1 Introduction of primary	
III	intermediates, unit processes	
	3.1.2 (a) Nitration	
	(b) Sulphonation	
	(0) Surpholiation	<u> </u>

(c) Halogenation (d) Diazotization : 3 different methods, importance (e) Ammonolysis (f) Oxidation N.B.: Definition, Reagents	
methods, importance (e) Ammonolysis (f) Oxidation	
(e) Ammonolysis (f) Oxidation	
(f) Oxidation	
N.B.: Definition, Reagents	
N.D.: Definition, Reagents	
Examples with reaction conditions	
(mechanism is not expected)	
3.2 Preparation of the following	
Intermediates.	
3.2.1 Benzene derivatives:	
Benzenesulphonic acid; 1,3-	
Benzenedisulphonic acid; phenol;	
resorcinol; sulphanilic acid; o-,m-	
,p-chloronitrobenzenes; o-,m-,p-	
nitroanilines; o-,m-p- phenylene	
diamines; Naphthol ASG.	
3.2.2 Naphthalene derivatives: α,β-	
Naphthols; α,β-Naphthylamines;	
Schaeffer acid, Tobias acid;	
Naphthionic acid; N.W. acid;	
Clev-6-acid; H acid; Naphthol As.	
3.2.3 Anthracene derivatives:	
1-Nitroanthraquinone;	
1-Aminoanthraquinone;	
2-Aminoanthraquinone;	
2-Methylanthraquinone;	
anthraquinone-1-sulphonic acid;	
Anthraquinnone-2-sulphonic acid;	
1-Chloroanthraquinone;	
Chloroanthraquinone;	
Benzanthrone.	
3.3 Dyeing Method of Cotton	
Fibres (3L)	
3.3.1 (i) Direct dyeing (ii) Vat	
dyeing (iii) Mordant dyeing	
(iv) Disperse dyeing	
3.3.2 Forces binding of dyes to the	
fibres: Ionic forces, Hydrogen	
bonds, Van-der-Wall's forces,	
Covalent linkages.	
4.1 Synthesis of Specific Dyes	
and their Uses (12L)	
IV (i) Orange IV from sulphanilic	
acid	
(ii) Eriochrome Black T from β-	

naphthol	
(iii) Eriochrome Red B by using	
ethyl aceto acetate and 1-amino-2-	
naphthol-4-sulphonic Acid.	
(iv) Direct Deep Black EW by	
using benzidine, H acid, aniline,	
and m-phenylen diamine.	
(v) Congo Red from nitrobenzene	
vi) Diamond Black F by using 5-	
amino salicylic acid, N.W. acid	
and α-naphthylamine.	
(vii) Malachite Green by using	
benzaldehyde and N,N-	
dimethylaniline.	
(viii) Auramine O from	
dimethylaniline	
(ix) Methylene Blue by using 4-	
amino-N,N-dimethylaniline and	
N,N-dimethylaniline	
(x) Safranine T by using o-	
toluidine and aniline	
(xi) Pararosaniline by using p-	
toluidine and aniline	
(xii) Alizarine Cyanine Green G	
by using phthalic anhydride and p-	
cholorophenol	
(xiii) Indanthrene from	
anthraquinone	
(xiv) Disperse Yellow 6G from	
benzanthrone	
(xv) Indigo from aniline	
(xvi) Eosine by using phthalic	
anhydride and resorcinol	
(xvii) Bismark Brown from m-	
phenylenediamine.	
4.2 Types of Fibres and Classes	
of Dyes Applicable to them (1L)	
Introduction to the following types	
of fibres with structures and	
classes of dyes applicable to it.	
Cotton, Wool, Silk, Polyester.	
4.3 Ecology and Toxicity of Dyes	
(2L)	
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With reference to the textile dyes, food colours, benzidine etc.	
1000 Colours, ochizianne etc.	

Practicals

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USACDD6P1	 I) Drug Preparation: Preparation of Aspirin from Salicylic Acid Drug Estimation: Estimation of Tincture of Iodine Estimation of Free Acid in Vegetable oil Dyeing of fabric (cotton) by Direct Dyeing or by Vat Dyeing 		
	 IV)Dyes Preparations: 1) Preparation of m-dinitrobenzene 2)) Preparation of m-nitroaniline V) Dye Estimation: 1.Estimation of Methyl Orange/ Eriochrome Black T/Eosin/Congo Red by colorimetry 	2	4

Modality of Assessment:

Theory Examination Pattern:

A) Internal Assessment - 40% 40 marks.

Theory 40 marks

Sr No	Evaluation type	Marks
1	One Assignments/Case study/Project	10
2	One class Test (multiple choice questions / objective)	20
3	Active participation in routine class instructional deliveries(case	05

	studies/ seminars//presentation)	
4	Overall conduct as a responsible student, manners, skill in articulation, leadership qualities demonstrated through organizing co-curricular activities, etc.	05

B) External examination - 60 %

Semester End Theory Assessment - 60%

60 marks

- i. Duration These examinations shall be of two hours duration.
- ii. Theory question paper pattern:-
- 1. There shall be **five** questions each of **12** marks. On each unit there will be one question & fifth one will be based on all the four units.
- 2. All questions shall be compulsory with internal choice within the questions. Each question will be of **24** marks with options.
- 3. Questions may be sub divided into sub questions a, b, c & d only, each carrying six marks **OR** a, b, c, d,e & f only each carrying four marks and the allocation of marks depends on the weightage of the topic.

Practical Examination Pattern:

(A)Internal Examination:-

There will not be any internal examination/ evaluation for practicals.

(B) External (Semester end practical examination) :-

Sr.No.	Particulars	Marks
1.	Laboratory work	80
2.	Journal	10
3.	Viva	10

Assessment pattern for semester end / External practical examination of 80 marks shall be finalized in the workshop of the subject

Semester end practical examination in applied component shall be conducted by the concerned department of the Institute/ College at the end of each semester and the marks of the candidates are to be sent to the University in the prescribed format.

Semester V:

Practical examination will be held at the college / institution at the end of the semester.

The students are required to present a duly certified journal for appearing at the practical examination, failing which they will not be allowed to appear for the examination.

In case of loss of Journal and/ or Report, a Lost Certificate should be obtained from Head of the Department/ Co-ordinator of the department; failing which the student will not be allowed to appear for the practical examination.

Semester VI

Practical examination will be held at the college / institution at the end of the semester. The students are required to present a duly certified journal for appearing at the practical examination, failing which they will not be allowed to appear for the examination.

In case of loss of Journal and/ or Report, a Lost Certificate should be obtained from Head of the Department/ Co-ordinator of the department; failing which the student will not be allowed to appear for the practical examination.